

DEVELOPMENT OF MWCNT EMBEDDED MICROMECHANICAL RESONATOR WORKING AS RAREFIED GAS SENSOR



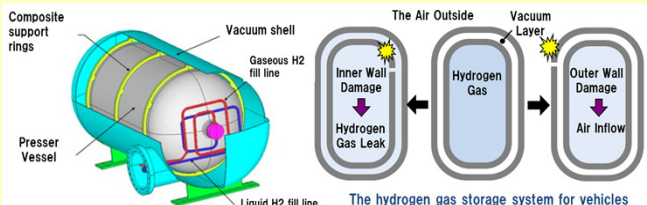
Hiroyuki KISHIHARA¹, Itsuo HANASAKI², Naoki MATSUDUKA³, Ichiro YAMASHITA⁴, Yukiharu URAOAKA⁴, and Yoshitada ISONO¹

¹Kobe University, ²Osaka University, ³Akashi National College of Technology, and ⁴Nara Institute of Science and Technology, JAPAN

ABSTRACT : This research has newly developed the multi-wall carbon nanotubes (MWCNTs) embedded-micromechanical resonator working as a novel rarefied gas sensor. The inertial effect of rarefied gas fluid is detected as a variation of the resonance frequency, and the dissipation of the interaction energy between the resonator and the gas molecules affects the damping of oscillation. Thus, two kinds of gaseous species can be distinguished with one device. The MWCNTs have been arranged on the resonator for heightening its sensitivity by the bio-MEMS compatible process. The MWCNTs embedded-resonator has successfully demonstrated to detect and distinguish hydrogen and nitrogen gases under pressures of 0.02 Pa to 0.9 Pa.

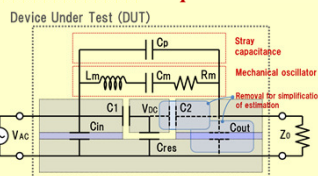
Design And Process of MWCNTs Embedded Micromechanical Resonator

Concept of Rarefied Gas Sensing



The objective of this research is to develop a novel rarefied gas sensor for a hydrogen gas storage system of vehicles. The MWCNTs embedded-micromechanical resonator can detect the inertial effect of rarefied gas fluid and the dissipation of the interaction energy between the resonator and the gas molecules.

Equivalent LCR circuit model of Resonator



$$S_{21} = \frac{2Z_0 Z_A (Z_m + Z_p)}{Z_0 (Z_0 + 2Z_A) (Z_m + Z_p) + Z_m Z_p (Z_0 + Z_A)} \quad (3)$$

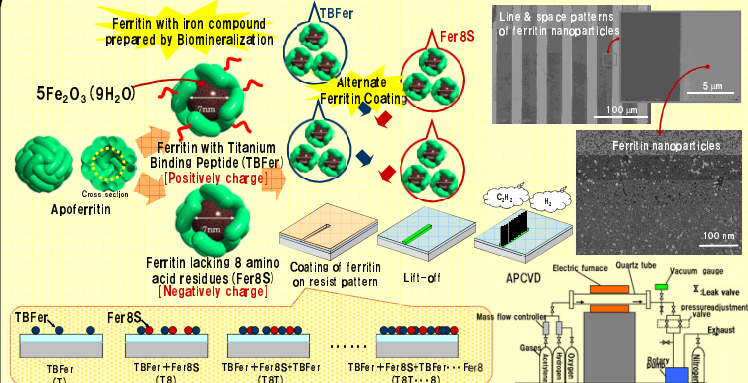
where,

$$Z_m = R_m + j\omega L_m + \frac{1}{j\omega C_m}, \quad Z_A = \frac{1}{j\omega C_A}$$

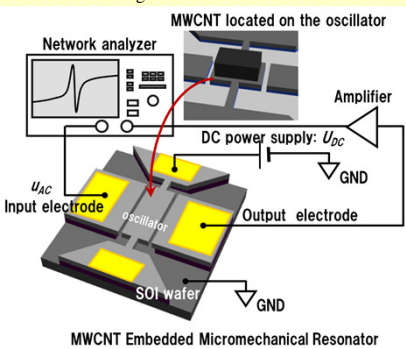
$$Z_p = \frac{1}{j\omega C_p}, \quad \text{and } C_A = \frac{C_1 C_{res}}{C_1 + C_{res}} + C_{in}$$

The equation of motion for the lumped mechanical model can be transferred into a simplified equivalent LCR circuit model. We have predicted the transmission amplitude $|S_{21}|$ as an output signal from the device using the Eq. (3) that is based on the LCR circuit model.

Bio-MEMS Compatible Process for MWCNTs Synthesis



In order to arrange MWCNTs on the resonator, this study employed MWCNTs synthesis using the charge-controlled ferritin proteins coating process for APCVD. Ferritin protein is a 12-nm-sized protein molecule with a hollow structure of 7 nm that can contain a $5Fe_2O_3 \cdot 9H_2O$ nanoparticle. In this study, the iron composite nanoparticles inside ferritin molecules were used as catalysts to synthesize MWCNTs in APCVD.



Considering the case of a Newtonian fluid, the fluid force on the oscillating resonator is described in the following equation

$$R(t) = -m_i \gamma_f u - m_i \beta_f \frac{du}{dt} \quad (2)$$

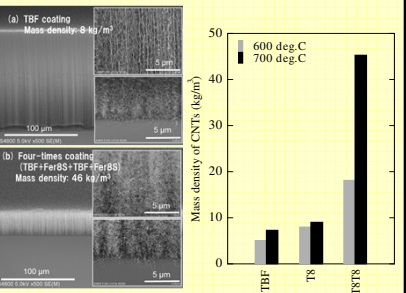
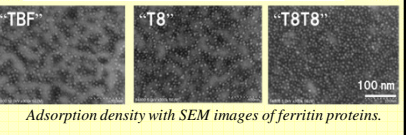
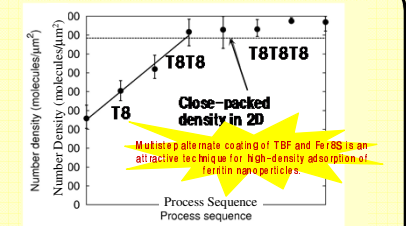
Rewriting Eq. (1) by the addition of Eq. (2), the following equation is derived.

$$(1 + \beta_f) \ddot{x} + (\gamma_f + \gamma_f) \dot{x} + \omega_0^2 x = F(t) / m_i$$

Therefore, the fluid can contribute to increases in both the effective mass and the dissipation of resonator.

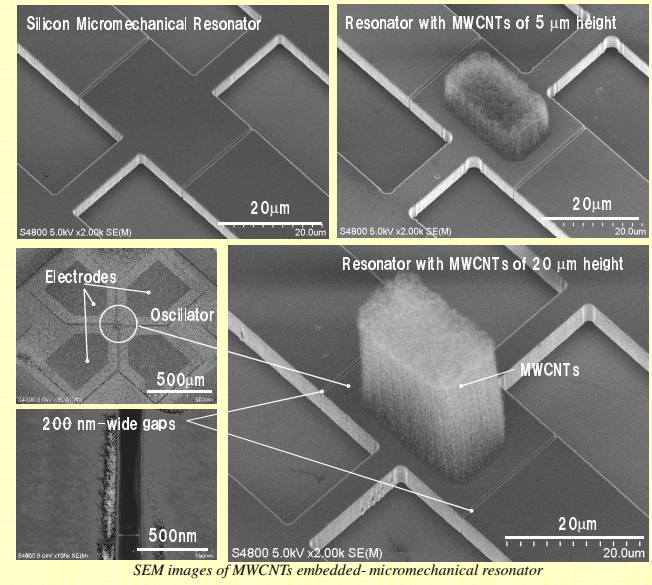
Results and Discussions

MWCNTs Synthesis



SEM images of MWCNTs synthesized from the ferritin protein catalysts: (a) TBF adsorption, and (b) multistep alternate adsorption of TBF and Fer8S. The density of the MWCNTs can be increased by an increase of ferritin density.

Characteristics of MWCNTs Embedded Micromechanical Resonator



The transmission amplitude $|S_{21}|$ of the device was measured at the U_{DC} of 50 V and the u_{ac} of 0 dBm.

- The anti-resonance frequency ratios in H_2 and N_2 gases increase with an increase of the pressure. This is caused by the increase of the effective mass due to inertial effect of the gas fluid around the resonator. The ratio in N_2 gas also shows the larger change than that in H_2 gas under each pressure because of the difference between their molecular weight.
- The similar trend of the damping parameter $1/\xi$ is observed. The change of $1/\xi$ in N_2 gas is larger than that in H_2 , which also depends on the molecular weight. From these figures, we can distinguish gaseous species of H_2 and N_2 .

